

Chemistry Minor
Paper: 1
(Theory)
Course Title: Chemistry in Daily life

Credits:4		Elective
Duration of the Course: 6 Months Total No. of Lectures = 60h		Max. Marks : 100
Unit	Topics	No. of Lectures
I	Chemistry of Water: Water resource, water quality parameter and drinking standard. Physical, chemical, biological quality of drinking water	06
II	Water composition analysis: Hardness testing, pH, salinity, turbidity/TDS, conductivity testing, minerals.	06
III	Carbohydrates: Classification of carbohydrates, general properties of glucose and fructose. Biological importance of carbohydrates.	06
IV	Basics of food science: Basic concepts, nutrition and nutrients, classification of food, food group and food pyramids and balanced diet.	06
V	(a) Food Preservation: Definitions, objective, and principle of food preservation, different methods of food preservation. (b) Contamination of food: Physical and chemical (heavy metal and pesticide) contamination of food. Elementary idea of food additives (intentional and unintentional, antioxidant, sweeteners and colors).	12
VI	Food safety and standards: Food safety, food hazards of biological origins, food standards (ISI, Agmark, FPO, MPO, PFA and FSSAI).	06
VII	Medicine and health care: Definition and Classification of drugs. Analgesic (non-narcotic and narcotic drugs). Adverse effect of the representative therapeutic agents (only elementary idea) such as sulfonamide chloroquine, chlorumphenicol, aspirin, paracetamol and antacids.	12
VIII	Cosmetics: Introduction of cosmetic products, forms of cosmetics and their suitable examples. Preventive measures in their uses and applications.	06
<p>Suggested Readings:</p> <ol style="list-style-type: none"> 1. Water Chemistry by Vernon L. Snoeyink and Davis Jenkins; John Wiley & Sons. 2. Water Chemistry by P. L. Brezonik and William A. Arnold; Oxford University Press. 3. Water Chemistry by Stanley E. Manahan; CRC Press, Taylor & Francis Group. 4. Drinking Water Treatment for Developing Countries' by A. B. Pandit and J. Kishen Kumar. 5. Carbohydrate based Drug Discovery' by Chi – Huey, Wong; Wiley – VCH. 6. Carbohydrate Chemistry for Food Scientists, 3rd Edition; J. N. BeMiller, WP: Woodhead Publishing, Elsevier. <p>Suggested online links: http://heecontent.upsdc.gov.in/Home.aspx https://nptel.ac.in/courses/104/106/104106096/</p>		

Chemistry Minor
Paper – 2
(Theory)
Course Title: Chemistry and Society

Credits:4		Elective
Duration of the Course: 6 Months Total No. of Lectures = 60h		Max. Marks : 100
Unit	Topics	No. of Lectures
I	Importance of Chemistry in Society: Chemicals – good or bad. Chemicals verses life expectancy. Consumer chemistry, chemical literacy for life.	06
II	Chemistry of Soil Science: Definition of soil, concept of lithosphere, soil as natural body, soil components (air, water, Inorganic and Organic solids).	06
III	Chemistry in Agriculture and Plant Protection: Fertilizers (nitrogen, phosphorus and potassium based) and biomagnifications. Pesticide (introduction and examples): insecticide, fungicides, herbicide, rodenticide and molluscicide. Regulations, Control and specific use of pesticide.	12
IV	Soap and detergents: Soap, types of soap, synthetic detergents and their classification. Mechanism of cleaning action.	06
V	Energy and Chemistry: Fuels (fossil, petroleum, coals and natural gas). Energy (nuclear, solar, water, wind), energy from biomass and garbage, green energy, clean energy.	10
VI	Chemistry in Global Economy: The direct and indirect global impact of chemical industry and their contribution towards sustainable developments.	06
VII	Chemistry and Modern Warfare: Explosives, nuclear weapons, chemical weapons of mass destruction. Type of chemical warfare agents (nerve, blood, blister, choking), protection devices.	08
VIII	Chemistry in Crime and Law Enforcement: Introduction of forensic chemistry, Chemistry in mob control, drugs in sports, analytical tool in crime detection.	06
<p>Suggested Readings:</p> <ol style="list-style-type: none"> 1. The Surface Chemistry of Soils by Garrison Sposito, Oxford University Press. 2. Principles and Practice of Soil Science: The Soil as a Natural Resource, 4th Ed. by Robert E. White; Blackwell Publishing. 3. Environmental Chemistry of Soils by Murray B. McBride, Oxford University Press. 4. Fundamentals of Soil Science by Henry D. Foth, 8th Ed., John Wiley & Sons. 5. Encyclopedia of Soil Science by Ward Chesworth, Springer. 6. Biobased surfactants and detergents: Synthesis, properties and applications; Douglas G. Hayes, Dai Kitamoto, D. K. Y. Solaiman, R. D. Ashby; AOCSS Press. 7. Lotions, Soaps and Scents; Anne G. Lewis; Lerner Publishing Group. 8. Handbook on Soaps, Detergents and Acid Slurry, 3rd rev. Ed., NIIR Board, Asia Pacific Business Press Inc. 9. Food, Energy and Water: The Chemistry connection; Satinder Ahuja, Elsevier. 10. Sustainability and Environmental Impact of Renewable energy Sources, R, M, Harrison, R. E. Hester, Royal Society of Chemistry. 11. Industrial Chemistry for Advanced Students; Mark Anthony Benvenuto; De Gruyter. 12. Agents of War: A History of Chemicals and Biological Weapons; Edward M, Spiers; Reaktion Books. 13. Forensic Chemistry: Fundamentals and Applications; Jay Siegel, Wiley – Blackwell. 14. Forensic Laboratory Manual: Chemistry Matter and Change (Student Edition), McGraw-Hill. 15. Forensic Chemistry: David E. Newton, Facts on File. 16. Genetically Modified Athletes: Biomedical Ethics, Gene Doping and sports; Andy Miah; Routledge. <p>Suggested online links:</p> <p>http://heecontent.upsdc.gov.in/Home.aspx https://nptel.ac.in/courses/104/106/104106096/ https://nptel.ac.in/courses/104/106/104106096/ https://www2.chemistry.msu.edu/faculty/reusch/VirtTxtJml/intro1.htm https://nptel.ac.in/courses/104/103/104103071/#</p>		